



**PHYLLURINE, LEAF-OPENING SUBSTANCE OF  
A NYCTINASTIC PLANT, *PHYLLANTHUS URINARIA* L.**

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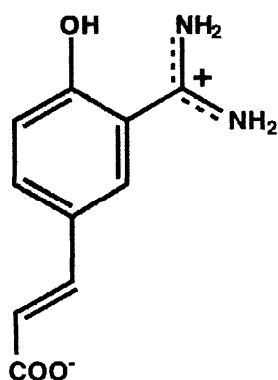
Received 24 September 1998; accepted 16 October 1998

**Abstract** : We have isolated phyllurine (**1**) as the leaf-opening substance of *Phyllanthus urinaria* L. that keeps the *Phyllanthus* leaves open. **1** was quite effective for the leaf-opening of *Phyllanthus urinaria* L. at  $2.5 \times 10^{-5}$  M, but not effective for other nyctinastic plants even at  $1 \times 10^{-4}$  M. The leaf-closing substance of this plant has already been identified as phyllanthurinolactone (**2**). Thus, the leaf-movement of *Phyllanthus urinaria* L. is proposed to be controlled by the interaction between **1** and **2**, similar to the case of *Lespedeza cuneata* G. Don. © 1998 Elsevier Science Ltd. All rights reserved.

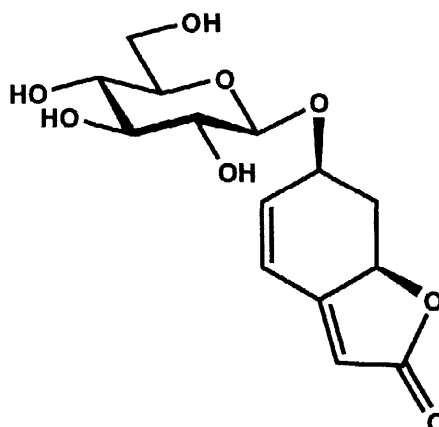
**Keywords**: plants; natural products; biologically active compounds; zwitter ions.

Most leguminosae plants close their leaves in the evening, as if to sleep, and open them in the morning.<sup>1</sup> This is called nyctinasty, and such a circadian rhythmic movement has been known to be controlled by their biological clocks.<sup>2</sup> Recently, we have identified several bioactive substances that regulate this leaf-movement,<sup>3</sup> and revealed that nyctinastic movement of the plants is controlled by the interaction between leaf-closing and -opening substances.<sup>4</sup> A biological clock controls the balance between these two bioactive substances.

Interestingly, a nyctinastic plant, *Phyllanthus urinaria* L., belongs to the *Euphorbiaceae* family. Because of this difference of species, the structure of phyllanthurinolactone (**2**),<sup>2c</sup> which is the only glycoside-type leaf-closing substance ever isolated, greatly differs from those from other leguminosae plants. Thus, it is important to identify the leaf-opening substance of this plant and reveal the differences in the regulating mechanism for nyctinastic movement.



**Phyllurine (1)**



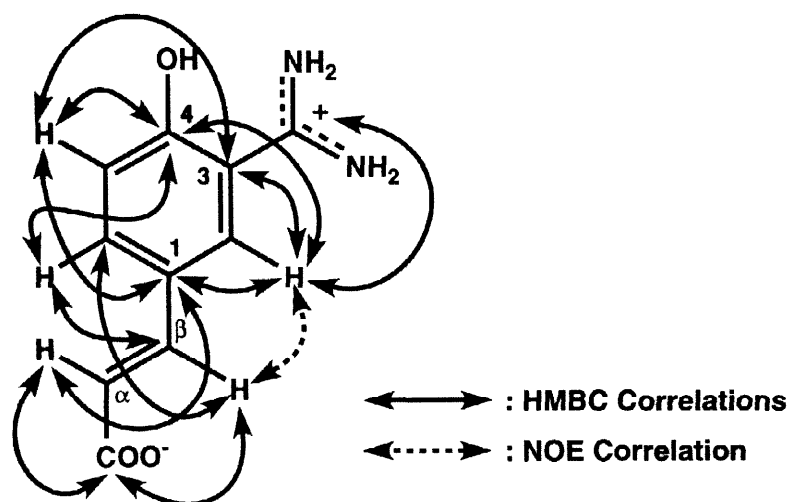
**Phyllanthurinolactone (2)**

Isolation of the leaf-opening substance was carried out based on a bioassay using a leaf of *Phyllanthus urinaria* L. The bioactive fraction keeps the leaves open until 6:00 PM.<sup>5</sup> The fresh whole plant of *Phyllanthus urinaria* L. (10.0 kg) was extracted with methanol for two weeks and concentrated *in vacuo*. The concentrated extract was partitioned with ethyl acetate, then with *n*-butanol. The bioactive aqueous layer was carefully separated by Amberlite XAD-7 column chromatography eluted with MeOH-H<sub>2</sub>O (0:10, 15:85, 25:75, 35:65, 50:50, and 10:0), and the 25%, 50%, and 100% MeOH aq. fractions showed weak leaf-opening activity. The 25% MeOH aq. fraction was analyzed, and it was revealed that this contained L-tryptophane as a leaf-opening substance. The 50% and 100% MeOH aq. fractions were further purified by MPLC using Develosil Lop ODS glass column with 20% MeOH aq., HPLC using preparative Develosil ODS HG-5 column with 20% MeOH aq., and HPLC using analytical Develosil ODS HG-5 column with 5% CH<sub>3</sub>CN aq. to give phyllurine (**1**, 0.5 mg).

L-Trp was effective at as low as  $1 \times 10^{-4}$  M on *Phyllanthus* leaves and the leaves of all other nyctinastic plants similar to indole-3-acetic acid (IAA). It is proposed that the bioactivity of L-Trp is attributed to IAA, which is known as an important metabolite of L-Trp.<sup>6</sup> IAA has been already reported to show weak leaf-opening activity to the leaves of all nyctinastic plants.<sup>7</sup> The bioassay was carried out by the addition of the sample solution at 11:00 PM.; thus, this long period necessary for the bioassay to detect the leaf-opening activity is sufficient for the metabolism of L-Trp into IAA.

On the other hand, **1** was effective at as high as  $2.5 \times 10^{-5}$  M only for the leaves of *Phyllanthus urinaria* L., and, similar to other leaf-opening substances, not effective for other nyctinastic plants, such as, *Aeschynomene indica* and *Albizia julibrissin* Durazz. even at  $1 \times 10^{-4}$  M. All of the leaf-movement factors previously isolated by us showed specific bioactivity on a plant species;<sup>3</sup> thus, the genuine leaf-opening substance of this plant should be **1**.

Structural determination of **1** was carried out by means of NMR and FAB MS experiments. HMQC, HMBC, and NOE experiments gave the structure of **1**.<sup>8</sup> The aromatic region of the <sup>1</sup>H NMR spectrum of **1** showed that **1** has a 1,2,4-trisubstituted aromatic ring. There was also observed a conjugated carboxylic acid moiety in this region. Correlations observed between these two parts gave the structure of *p*-coumarate substituted at the C<sub>3</sub>-position (Fig. 1). A strong molecular ion was observed in the positive-mode HR FAB-MS experiment to give the composition of **1** to be C<sub>10</sub>H<sub>11</sub>O<sub>3</sub>N<sub>2</sub>, while no molecular ion was observed in the negative-mode experiment. Thus, the residual part of **1** was deduced to be an amidinium ion, which is connected to the C<sub>3</sub>-position of **1**. This is supported by the weak correlation between H<sub>2</sub> and immonium carbon observed in HMBC experiment (Fig. 1). The presence of the amidine group was also supported by IR spectrum. A peak was observed at 1698 cm<sup>-1</sup> corresponding to the stretching of the imino group, together with a peak at 1603 cm<sup>-1</sup> corresponding to the stretching of the carboxylate group. This amidine carbon was difficult to detect by <sup>13</sup>C NMR experiment, probably because of the broadness of its signal, similar to the case of the guanidino function. The chemical shift of the C<sub>3</sub>-position in <sup>13</sup>C NMR spectrum shifted to lower field with the change of the solvent from CD<sub>3</sub>OD/D<sub>2</sub>O = 6/4 (129 ppm) into D<sub>2</sub>O (148.6 ppm).<sup>9</sup> This is attributed to the inhibition of the zwitter ionic structure of **1** in CD<sub>3</sub>OD/D<sub>2</sub>O = 6/4, whereas natural **1** is supposed to have zwitter ionic structure in the plant body.



**Fig. 1** Important Correlations Observed in HMBC and NOE Experiments.

Similar to other nyctinastic plants, the nyctinastic leaf-movement of *Phyllanthus urinaria* is assumed to be controlled by the competitive interaction between leaf-closing and -opening substances.<sup>4, 10, 11</sup> We have now isolated both of these substances, **1** and **2**, from *Phyllanthus urinaria*. Further studies on the regulation of the nyctinastic movement by a biological clock are now in progress. Our previous study on the nyctinastic movement of *Lespedeza cuneata* G. Don revealed that nyctinastic leaf-movement is controlled by a biological clock through the regulation of the activity of  $\beta$ -glucosidase which hydrolyzes the leaf-opening substance of this plant.<sup>4, 10</sup> A similar model would be applicable in the case of *Phyllanthus urinaria* L. In this case, the leaf-"closing" substance, which is a glucoside, would be hydrolyzed by the  $\beta$ -glucosidase whose activity is regulated by a biological clock. There is some possibility that the regulation of all nyctinastic leaf-movements can be explained by only one mechanism that either the leaf-closing or -opening substance is a glucoside in all nyctinastic plants. The biological clock regulates the activity of  $\beta$ -glucosidase which deactivates the glucoside to control the internal balance of concentration between leaf-closing and -opening substances.

**Acknowledgment** : We are indebted to the Ministry of Education, Science, Sports and Culture (Japan) for Grant-in-Aid for Scientific Research on Special Promote Research No. 09101001 and the Asahi Glass Foundation for financial support.

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5. Nyctinastic plants belonging to legminosae close their leaves around 6:00 PM. On the other hand, *Phyllanthus urinaria* L. closed their leaves around 3:00 PM. Leaf-opening activities are usually judged by the time lag of about three hours in the leaf-closing movement, thus, in this plant, we have judged the leaf-opening activity by the leaf-opening until 6:00 PM.
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8. Phyllurine (**1**):  $^1\text{H}$  NMR (400 MHz,  $\text{D}_2\text{O}$ , 35 °C): 7.75 (1 H, d,  $J = 16$  Hz,  $\text{H}_\alpha$ ), 7.25 (1 H, d,  $J = 2$  Hz,  $\text{H}_2$ ), 7.15 (1 H, dd,  $J = 2$  and 8 Hz,  $\text{H}_6$ ), 6.95 (1 H, d,  $J = 8$  Hz,  $\text{H}_5$ ), 6.50 (1 H, d,  $J = 16$  Hz,  $\text{H}_\beta$ ) ppm.;  $^{13}\text{C}$  NMR (100 MHz,  $\text{D}_2\text{O}$ , 35 °C): 170.5 ( $\text{C}_{\text{carbonyl}}$ ), 148.6 ( $\text{C}_3$ ), 147.0 ( $\text{C}_\beta$ ), 147.0 ( $\text{C}_\alpha$ ), 146.1 ( $\text{C}_{\text{immonium}}$ ), 145.8 ( $\text{C}_4$ ), 128.7 ( $\text{C}_6$ ), 124.3 ( $\text{C}_1$ ), 117.9 ( $\text{C}_5$ ), 116.8 ( $\text{C}_2$ ), ppm.; UV-vis ( $\text{H}_2\text{O}$ )  $\lambda_{\text{max}}$  ( $\epsilon$ ) 324 (7200), 302 (6600), 212 (12000) nm.; IR  $\nu$ : 1698, 1603  $\text{cm}^{-1}$ ; HR FAB-MS (positive):  $[\text{M} + \text{H}]^+$  Found  $m/z$  207.0807,  $\text{C}_{10}\text{H}_{11}\text{O}_3\text{N}_2$  requires  $m/z$  207.0844.
9. The chemical shift of the  $\text{C}_4$ -position in  $^{13}\text{C}$  NMR spectrum also shifted to higher field with the change of the solvent from  $\text{CD}_3\text{OD}/\text{D}_2\text{O} = 6/4$  (151.5 ppm) to  $\text{D}_2\text{O}$  (145.8 ppm).
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